# Introduction to molecular excitation and radiative transfer

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- Methods to solve the SEE
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## Molecular excitation The two-level approximation

Following Quantum Physics, a photon with a given frequency interacting with a quantum system with multiple energy levels could excite levels with an energy distance different than the photon energy, although with a small probability. The *two-level approximation* assumes that this is not possible and a photon only can produce a transition if its energy matches up with the energy difference between two levels.



Temperatures



## Statistical Equilibrium





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## Statistical Equilibrium



#### : radiative, spontaneous

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## Statistical Equilibrium

# $\frac{dn_i}{dt} = 0 = \sum_{j>i} A_{ji}n_j - \sum_{j < i} A_{ij}n_i$ $+ \frac{4\pi}{c} \sum_{j \neq i} B_{ji}\overline{j}(\nu_{ji})n_j - \frac{4\pi}{c} \sum_{j \neq i} B_{ij}\overline{j}(\nu_{ij})n_i$ $+ \sum_{j \neq i} C_{ij}n_j - \sum_{j \neq i} C_{ij}n_j$

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## Statistical Equilibrium





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## Statistical Equilibrium



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## The equation



$$\begin{split} I_{\nu}(\mathbf{r} + d\mathbf{r}, \mathbf{u}) d\omega ds d\nu &= I_{\nu}(\mathbf{r}, \mathbf{u}) d\omega ds d\nu dl + dI_{\nu}(\mathbf{r}, \mathbf{u}) d\omega ds d\nu \\ &= I_{\nu}(\mathbf{r}, \mathbf{u}) d\omega ds d\nu dl - \mathscr{A}_{\nu}^{g}(\mathbf{r}, \mathbf{u}) - \mathscr{A}_{\nu}^{d}(\mathbf{r}, \mathbf{u}) + \mathscr{E}_{\nu}^{g}(\mathbf{r}, \mathbf{u}) \\ &+ \mathscr{E}_{\nu}^{d}(\mathbf{r}, \mathbf{u}) + \mathscr{F}_{\star \to \nu}^{g}(\mathbf{r}, \star \to \mathbf{u}) + \mathscr{F}_{\star \to \nu}^{d}(\mathbf{r}, \star \to \mathbf{u}) \\ &- \mathscr{F}_{\nu \to \star}^{g}(\mathbf{r}, \mathbf{u} \to \star) - \mathscr{F}_{\nu \to \star}^{d}(\mathbf{r}, \mathbf{u} \to \star) \end{split}$$

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## The equation



$$\begin{split} I_{\nu}(\mathbf{r} + d\mathbf{r}, \mathbf{u}) d\omega ds d\nu &= I_{\nu}(\mathbf{r}, \mathbf{u}) d\omega ds d\nu dl + dI_{\nu}(\mathbf{r}, \mathbf{u}) d\omega ds d\nu \\ &= I_{\nu}(\mathbf{r}, \mathbf{u}) d\omega ds d\nu dl - \mathscr{A}_{\nu}^{g}(\mathbf{r}, \mathbf{u}) - \mathscr{A}_{\nu}^{d}(\mathbf{r}, \mathbf{u}) + \mathscr{E}_{\nu}^{g}(\mathbf{r}, \mathbf{u}) \\ &+ \mathscr{E}_{\nu}^{d}(\mathbf{r}, \mathbf{u}) + \mathscr{S}_{\star \to \nu}^{g}(\mathbf{r}, \star \to \mathbf{u}) + \mathscr{S}_{\star \to \nu}^{d}(\mathbf{r}, \star \to \mathbf{u}) \\ &- \mathscr{S}_{\nu \to \star}^{g}(\mathbf{r}, \mathbf{u} \to \star) - \mathscr{S}_{\nu \to \star}^{d}(\mathbf{r}, \mathbf{u} \to \star) \end{split}$$

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$$\begin{aligned} dI_{\nu}(\mathbf{r},\mathbf{u})d\omega ds d\nu &= -\mathscr{A}_{\nu}^{g}(\mathbf{r},\mathbf{u}) - \mathscr{A}_{\nu}^{d}(\mathbf{r},\mathbf{u}) + \mathscr{E}_{\nu}^{g}(\mathbf{r},\mathbf{u}) \\ &+ \mathscr{E}_{\nu}^{d}(\mathbf{r},\mathbf{u}) + \mathscr{S}_{\star \to \nu}^{g}(\mathbf{r},\star \to \mathbf{u}) + \mathscr{S}_{\star \to \nu}^{d}(\mathbf{r},\star \to \mathbf{u}) \\ &- \mathscr{S}_{\nu \to \star}^{g}(\mathbf{r},\mathbf{u} \to \star) - \mathscr{S}_{\nu \to \star}^{d}(\mathbf{r},\mathbf{u} \to \star) \end{aligned}$$

- $\mathscr{S}^{g}_{\star \to \nu}(\mathbf{r}, \star \to \mathbf{u})$  and  $\mathscr{S}^{g}_{\nu \to \star}(\mathbf{r}, \mathbf{u} \to \star)$ : Rayleigh scattering, Compton effect, etc., can be neglected in the CSM and ISM. We assume *complete frequency redistribution* [4, 5, 6, 7].
- $\mathscr{S}^{d}_{\star \to \nu}(\mathbf{r}, \star \to \mathbf{u})$  and  $\mathscr{S}^{d}_{\nu \to \star}(\mathbf{r}, \mathbf{u} \to \star)$ : Scattering by dust grains is not important for wavelengths larger than  $\simeq 8 \ \mu m$ .

$$dI_{\nu}(\mathbf{r},\mathbf{u})d\omega ds d\nu \simeq -\mathscr{A}_{\nu}^{g}(\mathbf{r},\mathbf{u}) - \mathscr{A}_{\nu}^{d}(\mathbf{r},\mathbf{u}) + \mathscr{E}_{\nu}^{g}(\mathbf{r},\mathbf{u}) + \mathscr{E}_{\nu}^{d}(\mathbf{r},\mathbf{u})$$

## The equation



The colisional constants  $C_{ul}$  and  $C_{lu}$  modify the population of the molecular levels, affecting the emitted and absorbed intensity.



The equation

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If the observed environment comprises molecular gas and dust:

$$\frac{dI_{\nu}}{dl} = -\left[k_{\nu}^{g} + k_{\nu}^{d}\right]I_{\nu} + \varepsilon_{\nu}^{g} + \varepsilon_{\nu}^{d}.$$

The typical form of the radiation transfer equation is:



For a given molecule with an abundance *x* and *m* transitions:

absorption profile (normalized)

gas density  
partition degeneracy of function the upper level
$$k_{\nu}^{g} = \frac{c^{2}}{8\pi} \frac{n_{g}x}{Z} \sum_{i=1}^{m} \frac{g_{u,i}A_{i}}{\nu_{i}^{2}} \left( \frac{g_{0,i}n_{l,i}}{g_{l,i}n_{0,i}} \right) \left( 1 - \frac{g_{l,i}n_{u,i}}{g_{u,i}n_{l,i}} \right) \phi_{i}(\nu - \nu_{i})$$

Assuming local termodynamical equilibrium (LTE,  $T_{exc} = T_{K} = T$ ):

$$k_{\nu}^{g} = \frac{c^{2}}{8\pi} \frac{n_{g}x}{Z} \sum_{i=1}^{m} \frac{g_{u,i}A_{i}}{\nu_{i}^{2}} e^{-E_{l,i}/k_{B}T} \left(1 - e^{-h\nu_{i}/k_{B}T}\right) \phi_{i}(\nu - \nu_{i})$$

$$\int_0^\infty d\nu \phi_i(\nu-\nu_i) = 1 \quad \Rightarrow \quad \phi_i(\nu-\nu_i) = \frac{1}{\sigma_i \sqrt{\pi}} e^{-(\nu-\nu_i)^2/\sigma_i^2}$$

There are several on-line databases available to find molecular spectroscopic parameters: (mm/submm) MADEX, Splatalogue, The CDMS Catalog, JPL, (IR) HITRAN, GEISA, (IR/visible) ExoMol...

For a given molecule with an abundance *x* and *m* transitions:

absorption profile (normalized)

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gas density  

$$k_{\nu}^{g} = \frac{c^{2}}{8\pi} \frac{n_{g}x}{Z} \sum_{i=1}^{m} \frac{g_{u,i}A_{i}}{\nu_{i}^{2}} \left( \frac{g_{0,i}n_{l,i}}{g_{l,i}n_{0,i}} \right) \left( 1 - \frac{g_{l,i}n_{u,i}}{g_{u,i}n_{l,i}} \right) \phi_{i}(\nu - \nu_{i})$$
partition  
function  
degeneracy of  
the upper level

The opacity of a given line is positive if

$$\zeta = 1 - \frac{g_l n_u}{g_u n_l} > 0 \Longrightarrow \frac{n_u}{n_l} < \frac{g_u}{g_l} \Longrightarrow T_{\text{exc}} > 0,$$

resulting in thermal emission, and negative if

$$\zeta = 1 - \frac{g_l n_u}{g_u n_l} < 0 \Longrightarrow \frac{n_u}{n_l} > \frac{g_u}{g_l} \Longrightarrow \frac{T_{\text{exc}}}{q_l} < 0,$$

producing maser emission (population inversion).

 $S_{\nu}$  of a line

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For a given molecule with an abundance *x* and *m* transitions:

$$S_{\nu}^{g} = \frac{2h}{c^{2}} \sum_{i=1}^{m} \frac{k_{\nu,i}^{g}}{k_{\nu}^{g}} \nu_{i}^{3} \left[ \frac{g_{u,i}n_{l,i}}{g_{l,i}n_{u,i}} - 1 \right]^{-1}$$
gas opacity
per line
total gas opacity

Assuming LTE ( $T_{exc} = T_{K} = T$ ):

$$S_{\nu}^{g} = \frac{2h}{c^{2}} \sum_{i=1}^{m} \frac{k_{\nu,i}^{g}}{k_{\nu}^{g}} \nu_{i}^{3} \frac{1}{e^{h\nu_{i}/k_{B}T} - 1} = \sum_{i=1}^{m} \frac{k_{\nu,i}^{g}}{k_{\nu}^{g}} B_{\nu_{i}}(T)$$
$$B_{\nu}(T) = \frac{2h\nu^{3}}{c^{2}} \frac{1}{e^{h\nu/k_{B}T} - 1} \qquad : \text{Planck's function}$$

The process followed to derive  $k_{\nu}^{g}$  and  $S_{\nu}^{g}$  can be strightforwardly extended to include other molecules.

For a solid state material which accounts for a fraction *y* of each grain:



Optical properties (refraction index: m = n + ik) [9, 10, 11, 12, 13, 14]



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For a solid state material which accounts for a fraction *y* of each grain:

$$k_{\nu}^{d} = n_{d}y\sigma_{\nu,i} = n_{d}y\sigma_{g}Q_{abs,i}(\lambda)$$

$$= n_{d}y\sigma_{g}\frac{\tilde{Q}_{abs,i}(\lambda)}{\lambda}$$

$$= n_{d}y\sigma_{g}\frac{1}{\lambda}\left\{-8\pi r_{g}\mathrm{Im}\left[\frac{m^{2}-1}{m^{2}+1}\right]\right\}$$
geometrical section
$$Thermal equilibrium + Gray body (S_{\nu}^{d} = B_{\nu}(T_{d}))$$

$$Total absorption = Total emission + \varepsilon_{\nu} \simeq k_{\nu}^{d}B_{\nu}(T_{d})$$

$$\int_{0}^{\infty} d\nu k_{\nu}^{d} \int_{4\pi} d\omega I_{\nu}(\omega) = 4\pi \int_{0}^{\infty} d\nu \varepsilon_{\nu}^{d} \simeq 4\pi \int_{0}^{\infty} d\nu k_{\nu}^{d}B_{\nu}(T_{d}) \implies T_{d}$$

The solution of the radiation transfer equation for a plane-parallel layer is:

$$I_{\nu} = I_{\nu,0} e^{-\tau_{\nu}} + e^{-\tau_{\nu}} \int_{0}^{\tau_{\nu}} dt S_{\nu}(t) e^{t}.$$

For a layer with constant temperature and density,  $S_{\nu} = \text{const.}$ :

$$I_{\nu} = I_{\nu,0} e^{-\tau_{\nu}} + S_{\nu}(T) \left(1 - e^{-\tau_{\nu}}\right).$$

The numerical solution of the radiation transfer equation is based on the discretization of the target environment in *m* layers with constant physical quantities:

$$I_{\nu} = I_{\nu,0} e^{-\sum_{j=1}^{m} \tau_{\nu,j}} + \sum_{j=1}^{m} S_{\nu}(T_j) \left(1 - e^{-\tau_{\nu,j}}\right) e^{-\sum_{p=1}^{j-1} \tau_{\nu,p}}. \quad j \leftarrow 0.54321$$

 $I_{\nu,0} \xrightarrow{\tau_{\nu}(l), S_{\nu}(l)} I_{\nu}$  T(l), n(l)  $T(l) \xrightarrow{T(l), n(l)} I$ 

$$I_{\nu,0} \quad \tau_{\nu}, \ S_{\nu} \neq f(l) \quad I_{\nu}$$

$$T, \ n \neq f(l)$$

 $I_{\nu,0}$ 

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## **Optical** regimes

$$I_{\nu} = I_{\nu,0}e^{-\tau_{\nu}} + S_{\nu}(T)\left(1 - e^{-\tau_{\nu}}\right) \quad ; \quad F_{\nu} = \int_{4\pi} P(\phi,\theta)I_{\nu}(\phi,\theta)d\omega \simeq \langle I_{\nu}\rangle\Omega$$

Optically thin regime

$$\begin{aligned} \tau_{\nu} \ll 1 \\ I_{\nu} \simeq I_{\nu,0} + \left[ S_{\nu}(T) - I_{\nu,0} \right] \tau_{\nu} \\ F_{\nu} \simeq \langle I_{\nu,0} + \left[ S_{\nu}(T) - I_{\nu,0} \right] \tau_{\nu} \rangle \Omega \\ \Omega = \text{solid angle} \end{aligned}$$

$$F_{\nu} = \int_{4\pi} P(\phi, \theta) I_{\nu}(\phi, \theta) d\omega \simeq \langle I_{\nu} \rangle \Omega$$



Optically *thick* regime

 $\tau_{\nu} \gg 1$  $I_{\nu} \simeq S_{\nu}(T)$  $F_{\nu} \simeq \langle S_{\nu}(T) \rangle \Omega$ 



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Assuming we have observed a set of rotational lines of a molecule and they are optically thin [15]:

$$\begin{split} F_{\nu}^{\text{observed}} \simeq \langle I_{\nu} \rangle \frac{\pi}{4} \left( \theta_{s}^{2} + \theta_{b}^{2} \right) &= \frac{2k_{B}\nu^{2}T_{mb}}{c^{2}} \frac{\pi}{4} \left( \theta_{s}^{2} + \theta_{b}^{2} \right) \quad , \quad T_{mb} = \frac{F_{\text{eff}}}{B_{\text{eff}}} T_{\text{A}}^{*} \\ F_{\nu}^{\text{emitted}} \simeq \frac{2k_{B}\nu^{2}T_{B}}{c^{2}} \frac{\pi}{4} \theta_{s}^{2} &= \frac{h\nu}{4\pi} g_{u}A_{ul} \frac{N_{\text{col}}}{Z} e^{-E_{u}/k_{B}T_{\text{rot}}} \phi_{\nu} \frac{\pi}{4} \theta_{s}^{2} \\ W = \int_{0}^{\infty} T_{mb} dv \simeq \frac{hc^{3}g_{u}A_{ul}}{8\pi k_{B}\nu^{2}} \frac{N_{\text{col}}}{Z} e^{-E_{u}/k_{B}T_{\text{rot}}} \frac{\theta_{s}^{2}}{\theta_{s}^{2} + \theta_{b}^{2}} \\ \ln\left(\frac{8\pi k_{B}\nu^{2}W}{hc^{3}g_{u}A_{ul}}\right) \simeq \ln\left(\frac{N_{\text{col}}}{Z} \frac{\theta_{s}^{2}}{\theta_{s}^{2} + \theta_{b}^{2}}\right) - \frac{E_{u}}{k_{B}T_{\text{rot}}} \end{split}$$

In the optically thick regime, the additional quantity  $\ln [(1 - e^{-\tau})/\tau]$  needs to be added to the right hand term. escape probability ( $\beta$ ][20, 21, 22, 23, 3, 24]

## Rotational diagrams

Assuming we have observed a set of rotational lines of a molecule and they are optically thin [26]:



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## Line profiles

Vexp

 $v_{\rm sys}$ 

## Light Doppler effect

**Relativistic** Doppler effect



Non-relativistic Doppler effect

 $\frac{\nu - \nu_0}{\nu_0} \simeq -\frac{v}{c}$ 



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## Line profiles

## Turbulence and gas kinematics



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- Accelerated  $\Lambda$  iteration (ALI) [17, 30]: It uses the  $\Lambda_{ij}$  operator to calculate iteratively the total intensity and the source function at any point of the modeled environment.
- Large Velocity Gradient (LVG; Sobolev method) [22, 25]: A photon can escape from the region where it has been emitted due to differences in velocity with other close regions. It is the ALI method with  $\Lambda_{ij} = 1 \beta$ .
- Monte Carlo [18, 28]: A number of model photons (comprising many "real" photons) related to an initial source function travel through the modeled environment modifying the population of the molecular levels. The process is repeated to reach convergence.
- Gauss-Seidel algorithm [31, 32, 16]: The population of the molecular levels in a shell are recalculated once the total intensity for this shell is known. It is not necessary to solve the SSE for all the environment at the same time.
- Coupled escape probability (CEP) [34]: The molecular level populations are calculated without knowing the total radiation intensity in every point of the environment to model.

#### Numerical codes

### Typical resulting lines



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